



COURSE DESCRIPTION CARD - SYLLABUS

Course name

Technologia chemiczna organiczna

Course

Field of study

Technologia Chemiczna (Chemical Technology)

Area of study (specialization)

Level of study

First-cycle studies

Form of study

full-time

Year/Semester

III/6

Profile of study

general academic

Course offered in

Polish

Requirements

compulsory

Number of hours

Lecture

30

Laboratory classes

30

Other (e.g. online)

Tutorials

15

Projects/seminars

Number of credit points

6

Lecturers

Responsible for the course/lecturer:

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Faculty of Chemical Technology

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Responsible for the course/lecturer:

Prerequisites

Student has knowledge of general, organic and inorganic chemistry, knows basic methods, techniques and tools used in chemical analysis.

Student is able to obtain information from literature, databases and other sources, is able to interpret obtained information, draw conclusions and formulate opinions.

Student is able to apply the acquired knowledge in practice, both during his professional work and during further education.

Student is able to cooperate and work in a group.

Student is able to adequately determine priorities for the realization of a given task.



Course objective

Obtaining knowledge of organic chemical technology.

Course-related learning outcomes

Knowledge

1. Student has knowledge of raw materials, products and processes used in the chemical industry and of the directions of development of the chemical industry in the country and worldwide [K_W09]
2. Student has a structured general and detailed knowledge of chemical technology [K_W13]
3. Student is familiar with basic methods, techniques, tools and materials used to solve simple organic chemical technology tasks. [K_W15]

Skills

1. Student has the ability to self-study. [K_U05]
2. Student explains, on the basis of general knowledge, basic phenomena related to important processes in chemical technology and engineering (K_U16)

Social competences

1. Student understands the need for further education and improvement of his professional and personal competences. [K_K03]
2. Student is aware of the responsibility for his own work and willingness to submit to teamwork and take responsibility for jointly performed tasks. [K_K04]

Methods for verifying learning outcomes and assessment criteria

Learning outcomes presented above are verified as follows:

Lecture - written examination; evaluation criterium: 3 - 50.1-70.0%; 4 - 70.1-90.0%; 5 - from 90.1%

Laboratory: current control during laboratory classes, oral/written response, reports of laboratory exercises, oral/ written response, evaluation of teamwork; evaluation criterium: 3 - basic theoretical and practical preparation, the ability to prepare reports from laboratory exercises; 4 - practical preparation supported by theoretical knowledge, the ability to formulate appropriate conclusions, active participation in classes supported by the desire to obtain additional knowledge; 5 - complete preparation for teaching classes, the ability to formulate conclusions at an advanced level, precise performance of the tasks entrusted, independent search for additional theoretical knowledge, coordination of work in a research team.

Exercises - colloquia during classes

Programme content

1. Energy sources (from coal to nuclear energy).
2. Technological principles (the principle of potential difference, best use of raw material, best use of energy, best use of equipment, technological moderation). The principle of patent purity.



3. Chlorination process (basic reactions, chlorinating agents, organic chemistry, thermodynamics and kinetics, examples, waste and its disposal, technological schemes).
4. Alkylation process (reactions, alkylating agents, Friedel-Crafts process, high octane gasoline, waste and its utilization, technological schemes).
5. Neutralization technology.
6. Biomass as a chemical raw material (oils and fats, lignocellulose products, soaps, fatty acids, glycerine, biological activity products, technological schemes).
7. Ionic liquids (synthesis, properties, application, disposal, green solvents).

Teaching methods

Lecture - multimedia presentation, illustrated with examples given on the board.

Laboratory - educational materials for the laboratory in the form of pdf files, practical exercises.

Exercises - calculation tasks solving from organic chemical technology

Bibliography

Basic

1. E. Grzywa, J. Molenda: Technologia podstawowych syntez organicznych, T. 1 i 2, WNT, Warszawa 2008.
2. E. Kociotek-Balawejder (red.): Technologia chemiczna organiczna: wybrane zagadnienia, Wydawnictwo Uniwersytetu Ekonomicznego we Wrocławiu, 2013.
3. M. Taniewski: Technologia chemiczna - surowce, Wydawnictwo Politechniki Śląskiej, Gliwice 1997.
4. M. Stasiewicz (red.): Technologia chemiczna organiczna, ćwiczenia laboratoryjne, Wydawnictwo Politechniki Poznańskiej, Poznań, 2013.
5. B. Burczyk: Biomasa. Surowiec do syntez chemicznych i produkcji paliw, Oficyna Wydawnicza Politechniki Wrocławskiej, Wrocław 2011.
6. B. Burczyk: Zielona chemia. Zarys, Oficyna Wydawnicza Politechniki Wrocławskiej, Wrocław 2014.

Additional

1. J.A. Moulijn, M. Makkee, A. van Diepen: Chemical Process Technology, Wiley-Blackwell, Chichester 2013.
2. M. Taniewski: Przemysłowa synteza organiczna. Kierunki rozwoju, Wydawnictwo Politechniki Śląskiej, Gliwice 1991.
3. P. Wasserscheid, T. Welton: Ionic liquids in synthesis, Wiley-VCH, Weinheim 2003.



Breakdown of average student's workload

	Hours	ECTS
Total workload	150	6,0
Classes requiring direct contact with the teacher	85	3,4
Student's own work (literature studies, preparation for the laboratory classes, preparation for the exam, elaboration of research results and preparation of reports from the laboratory classes) ¹	65	2,6

¹ delete or add other activities as appropriate